From Silphinenes to Penifulvins: A Biomimetic Approach to Penifulvins B and C

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ABSTRACT

The biomimetic total synthesis of penifulvins B and C uses a *meta***-photocycloaddition as a key step which gives rapid access to the fenestranetype carboskeleton with control of an onring quaternary stereocenter. The route is concise, stereocontrolled, scaleable, and flexibile and requires only one protecting group.**

Recently, Gloer et al. described the isolation of novel sesquiterpenoids named penifulvin A, and B, C $(1-3)$, as well as 12-hydroxy-silphinen-15-oic acid **4**¹ from the fungus *Penicillium griseoful*V*um* (syn. *P. patulum* Bain.; *P. urticae* bain.), which were obtained from white mycelia growth on dead hardwood branches collected in Hawaiian forests. Among these metabolites, **1** shows significant antiinsectan activity in assays against the fall armyworm *Spodoptera frugiperda*² The overall structures of **1–3** and **4** were secured (except for the absolute configurations) by X-ray crystallographic analysis and revealed a highly complex dioxa[5.5.5.6]fenestrane structure in which four rings share a central quaternary carbon atom (Figure 1). **1** and **4** contain five stereocenters, whereas **2** and **3** have six of them, three of which are quaternary.

Figure 1. Structures of penifulvins A-C and 12-hydroxysilphinene-15-oic acid.

Due to this unusually dense array of stereogenic carbons, these compounds represent attractive targets for total synthesis. Recently, we disclosed the synthesis of penifulvin A (**1**).³ Herein we wish to report two enantioselective total syntheses of penifulvins B (**2**) and C (**3**) by means of a *meta*photocycloaddition^{4,5} (Scheme 1).

Our synthesis was guided by two major interests. First we wanted to corroborate our biosynthetic hypothesis for

^{(1) (}a) Penifulvin A: Shim, H. S.; Swenson, D. C.; Gloer, J. B; Dowd, P. F; Wicklow, D. T. *Org. Lett.* **2006**, *8*, 1225–1228. (b) Penifulvins B and C, 12-hydroxy-silphinen-15-oic acid: Shim, H. S.; Gloer, J. B.; Wicklow, D. T *J. Nat. Prod.* **2006**, *69*, 1601–1605. In this paper also, penifulvins D and E were described.

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the oxidative formation of the dioxa[5.5.5.6]fenestrane structure type (e.g., **3**) from the well-known silphinene carbon skeleton (4) (Scheme 2).⁶

As penifulvins B and C are epimers regarding the C-6 quaternary carbon center, they provide us with a second interesting opportunity, the testing of the suprafacial selectivity of the *meta*-photocycloaddition, which is the key transformation in our synthesis (Scheme 3). If we postulate that the photoreaction forms two σ -bonds (C1-C8 and C2-C6) to generate an exciplex E_1 first, the C1-C8 bond might be reopened in the next step to form a highly stabilized cyclohexadienyl-*tert*-alkyl-diradical species E_2 which undergoes 1,2-rotation to E_3 and recloses the C1-C8 bond to E_4 . It thus depends on the relative rates of $C7 - C9/C7 - C11$ ring closure (formation of a highly strained cyclopropyl system) vs C1-C8 ring opening (considerable strain release) whether regioisomers **9**/**10** alone or additionally **11**/**12** are obtained. Direct ring closures of E_2 and E_3 are unlikely due to high ring strain. Similarly, a $C1-C7$ or $C1-C11$ combination in E_2/E_3 would create a highly strained cyclobutane ring. Another issue is the equilibrium between the

two potentially reactive conformers 8a/8b. The A^{1,3}-strain⁷ between the 5-*R*-group and the 7-Me **8b** is strongly disfavored, so **8a** may be assumed as the preferred conformer, which directs the cycloaddition onto the top-face of the benzenoid ring.

We thus prepared the trisubstituted *E*/*Z*-isomers **15** (for **2**) and **5** (for **3**) via Myers alkylation⁸ (Scheme 4) and submitted them to photocyclization. To our delight, the photoreaction proceeded exclusively with suprafacial stereochemistry (Scheme 5). Hence **19** and **6** were formed from the irradiation of *Z*-isomer **5**, and **17** and **18** were formed from *E*-isomer **15**. Both cyclizations proceeded via *exo* transition states.⁹ The synthesis was completed by separation

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Scheme 5. *meta*-Photocycloaddition of **15** and **5**

of the regioisomeric pairs (**17/18** and **19/6**), which could be re-equilibrated via a thermal vinylcyclopropane-cyclopentene $rearrarrangement¹⁰$ so that no material was lost. Next the cyclopropane ring in **18** was opened regioselectively under Birch-type conditions, and the primary alcohol was oxidized to carboxylic acid **20** (Scheme 6). Ozonolytic cleavage of the double bond converted **20** via a nonisolable dialdehyde into the hemiacylal **21**. Further oxidation to lactone **22** and

removal of the protective group furnished penifulvin B (**2**). An analogous sequence was used to transform **6** into **3** (Scheme 7). All analytical data of our synthetic samples of **2** and **3** matched those reported by Gloer.1b

Although all intermediates could be isolated and characterized, their ultimate purification proved difficult and was thus postponed to **22**/**25**, which were shown to be diastereomerically pure with respect to the crucial quaternary center at C6.

In conclusion, we have developed the first enantioselective total syntheses of penifulvins B and C (**2** and **3**). Both syntheses, which confirm the absolute configurations tentatively assigned previously,^{1b} use only one protecting group and are concise, requiring 10 steps for **2** and 13 steps for **3**, respectively. The endgame supports our biosynthetic hypothesis, in which the silphinene carbon skeleton is converted into the dioxa[5.5.5.6]fenestrane ring system in a cascade oxidation. Additionally our findings exponentiate the synthetic value of the *meta*-photocycloaddition, because due to its suprafacial stereochemistry with respect to the olefin component the challenging problem of synthesizing stereogenic on-ring quaternary carbon centers¹¹ is reduced to selectively generating trisubstituted E/Z double bonds.¹²

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Supporting Information Available: Experimental data and analytical characterization for all new compounds provided. This material is available free of charge via the Internet at http://pubs.acs.org.

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